

sequences of the same which are known and readily available. As the terms have a well-established meaning in the art and the sequences corresponding to those genes are both known and can be obtained by one of skill in the art, the terms *ushA* gene and *aphA* gene are definite.

Concerning 5'-inosinic acid or 5'-guanylic acid, Applicants submit herewith and direct the Examiner's attention to select pages from the Merck Index and the entries corresponding to the same demonstrating that the terms 5'-inosinic acid or 5'-guanylic acid are normally used by one of skill in the art to define a nucleoside 5'-phosphate ester. This is also consistent with the description in the present specification found in the paragraph bridging pages 1 and 2.

In view of the foregoing, withdrawal of the rejections under 35 U.S.C. § 112, second paragraph is requested.

The rejection of Claims 4 and 5 under 35 U.S.C. § 102(b) over Laird et al is respectfully traversed.

Laird et al describes "E coli mutants incapable of *de novo* purine biosynthesis and also lacking other periplasmic enzymes with 5'-nucleotidase activity (*ushA* and *aphA*).” However, Laird et al do not describe an Escherichia bacteria with the *ushA* and *aphA* genes disrupted and which has an ability to produce and accumulate nucleoside 5'-phosphate esters in a medium. Therefore, Laird et al does not anticipate the present claims and as such withdrawal of this ground of rejection is requested.

The rejection of Claims 4 and 5 under 35 U.S.C. § 103(a) over Thaller et al alone or in view of Cowman et al is respectfully traversed.

Thaller et al describe the identification of the aphA gene and on page 197, second paragraph that the aphA gene is a "physiological equivalent to the ushA gene." Thaller et al further describe "characterization of the parameters of this enzyme toward selected substrates, along with investigations on strains carrying genetically defined aphA mutations, are warranted to understand the physiological role of this class of highly conserved bacterial enzymes and to ascertain the significance of the phosphotransferase activities shown by these enzymes under laboratory conditions" (see page 198, col. 1). Therefore, while Thaller et al may describe the potential usefulness of studying aphA by mutating the gene, Thaller et al does not describe the claimed bacterium which has both the ushA and aphA genes disrupted and which has an ability to produce and accumulate nucleoside 5'-phosphate esters in a medium. Cowman et al merely describes the cloning of the ushA gene but also does not describe the nucleoside 5-phosphate ester producing and accumulating property found when the ushA gene and aphA genes have been disrupted in the bacteria. Therefore, in combination, the cited prior art provides no description for the claimed invention.

As shown in Tables 6 and 7 on pages 35 and 37, respectively, disruption of both genes facilitated the production and accumulation of IMP and GMP in the medium. For the Examiner's reference Table 6 is reproduced below:

Strain	Culture time (h)	Inosine (g/L)	IMP (g/L)
I/pMWpurFKQ	48	2.3	0
	48	2.3	0
I Δ ushA/pMWpurFKQ	51	3.1	0
	51	2.9	0
I Δ aphA/pMWpurFKQ	51	3.6	0
	51	3.2	0
I Δ ushA Δ /aphA/pMWpurFKQ	54	2.4	1.0
	54	2.6	0.6

The data in this Table demonstrate that only the bacterial strain deficient in both genes (row 4) was able to produce and accumulate IMP and the medium compared to either gene mutant alone (rows 2 and 3) or the parental strain (row 1). Therefore, even if one assumes that it would have been obvious to disrupt both genes, there would not have been an expectation that disrupting both genes rather than each individually would facilitate the production of nucleoside 5'-phosphate esters. This is particularly so in light Thaller et al who describes the aphA gene is a physiological equivalent of ushA gene

Therefore, the present claims are not obvious in view of the combination of Thaller et al and Cowman et al. Withdrawal of this ground of rejection is requested.

Applicants submit the present application is ready for allowance. Early notification of such allowance is kindly requested.

Respectfully submitted,

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IN THE CLAIMS

4. (Amended) [A] An isolated bacterium belonging to the genus Escherichia having an ability to produce and accumulate nucleoside 5'-phosphate ester in a medium, in which ushA gene and aphA gene are disrupted.

5. (Amended) The isolated bacterium belonging to the genus Escherichia according to Claim 4, wherein the nucleoside 5'-phosphate ester is selected from the group consisting of 5'-inosinic acid or 5'-guanylic acid.

Claims 1-3 and 6-8 are canceled.

Claims 9 and 10 are added.

THE MERCK INDEX

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CHEMICALS, DRUGS, AND BIOLOGICALS

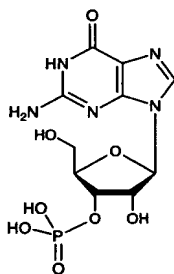
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$C_{10}H_{14}N_5O_8P$; mol wt 363.22. C 33.07%, H 3.88%, N 19.28%, O 35.24%, P 8.53%. From yeast or pancreas. Prepn: P. A. Levene, L. W. Bass, *Nucleic Acids* (New York, 1931) pp 224-227. Structure: Levene, Jorpes, *J. Biol. Chem.* 81, 579 (1929); Levene, Harris, *ibid.* 95, 755 (1932); 98, 9 (1932). Early work probably done on a mixture of 2'- and 3'-guanylic acids; see physical data below. Separation of two isomers: Cohn, *J. Am. Chem. Soc.* 72, 1471 (1950); Khym, Cohn, *ibid.* 76, 1818 (1954); *idem*, *Biol. Prepn.* 5, 40 (1957). Absorption spectrum: Voet *et al.*, *Biopolymers* 1, 193 (1963). Reviews: see Guanine, *Nucleic Acids*.

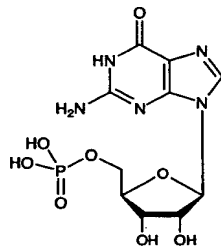


Dihydrate, long prisms from water. The water of crystn is given up at 118° and is taken up again at room temp. When anhydrous, dec 180° (closed tube). $[\alpha]_D^{25} -8^\circ$ (c = 2); -65° (c = 2 in 5% NaOH). Acid to litmus. Soluble in cold water, freely sol in hot water. Boiling with dil mineral acids yields guanine, H_3PO_4 , and D-ribose.

Neutral sodium salt, $Na_2C_{10}H_{12}N_5O_8P$, flakes from water, contains 21.1% H_2O . Sol in cold, freely sol in hot water.

Brucine salt heptahydrate, $C_{10}H_{14}N_5O_8P \cdot (C_{23}H_{28}N_2O_4)_2 \cdot 7H_2O$, rectangular leaflets from alc. When anhydr, dec 233-240°. $[\alpha]_D^{25} -26^\circ$ (35% alc). One gram dissolves in 100 ml water.

4600. 5'-Guanylic Acid. Guanosine 5'-monophosphate; GMP; guanosine 5'-phosphate; guanine riboside-5-phosphoric acid. $C_{10}H_{14}N_5O_8P$; mol wt 363.22. C 33.07%, H 3.88%, N 19.28%, O 35.24%, P 8.53%. Nucleotide widely distributed in nature; found in hydrolyzates of RNA. Isolated together with inosinic acid from sardines or yeast extract: Kuninaka *et al.*, *New Food Ind. (Tokyo)* 3, no. 1, 21 (1961). Also by direct biosynthesis using microorganisms or enzymes: Abrams, Bentley, *Arch. Biochem. Biophys.* 79, 91 (1959); Magasanik, Karibian, *J. Biol. Chem.* 235, 2672 (1960); Okumura *et al.*, U.S. pat. 3,249,511 (1966). Chemical synthesis: Michelson, Todd, *J. Chem. Soc.* 1949, 2483; Chambers *et al.*, *J. Am. Chem. Soc.* 79, 3747 (1957); Gilham, Tener, *Chem. & Ind. (London)* 1959, 542; Tener, *J. Am. Chem. Soc.* 83, 159 (1961); Koransky *et al.*, *Z. Naturforsch.* 17B, 291 (1962). Prepn of Na salt: Ishibashi, Ito, U.S. pat. 3,190,877 (1965 to Takeda). Monograph on synthesis of nucleotides: G. R. Pettit, *Synthetic Nucleotides* vol. 1 (Van Nostrand Reinhold, New York, 1972) 252 pp. Reviews: See Guanidine; *Nucleic Acids*.

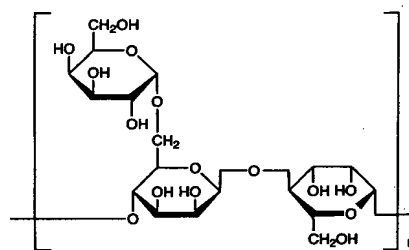


Microcrystals, dec 190-200°. Sparingly sol in cold water. Barium salt octahydrate, $C_{10}H_{12}N_5O_8P \cdot Ba \cdot 8H_2O$, white powder. uv max (pH 2): 256 nm (ϵ 12400); (pH 12): 260 nm (ϵ 12100).

Disodium salt monohydrate, $C_{10}H_{12}N_5O_8PNa_2 \cdot H_2O$, microscopic crystals, decomp at about 250°. Characteristic meaty taste. a_m (molar absorptancy): 13.7×10^3 at 256 nm (pH 7). Soly in water at 25° about 25 g/100 ml. Practically insol in alcohol, acetone, ether.

USE: The disodium salt as flavor intensifier, like sodium inosinate and sodium glutamate. Said to be more effective than either.

4601. Guar. Principal polysaccharide from seed of guar seeds, *Cyamopsis tetragonoloba* (L.) Desf. *Leguminosae*: Heyne, Whistler, *J. Am. Chem. Soc.* 70, 151 (1948). Structure: Whistler, Durso, *ibid.* 74, 5140 (1952). Configuration: Koleske, Kurath, *J. Polymer Sci. Polym. Chem. Ed.* 4, 123 (1966). Review: Deuel *et al.*, *Chimia* 8, 64 (1954).



$[\alpha]_D^{25} +53^\circ$ (1N NaOH). Sol in cold water. Triacetate, fibrous material, mp 226-227°. Can be formed into strong films which can be elongated 550%. Birefringent and does not develop crystallinity. USE: In textile and paper industry.

4602. Guar Gum. Guar flour; gum cyamopsis; mopsis gum; Burtonite V-7-E; Jaguar; Decarpa; Glucotard; Guarem. Mol wt about 220,000. The endosperms of *Cyamopsis tetragonoloba* (L.) Taubert, *Leguminosae* which is cultivated in India as livestock feed, water soluble fraction (85%) of guar flour is called guar gum which consists of linear chains of (1→4)-β-D-mannosyl units with α-D-galactopyranosyl units attached (1→6) linkages. Ratio of D-galactose to D-mannose = 1:2. Effect on lipid metabolism: D. J. A. Jenkins *et al.*, *Med. J.* 2, 1555 (1979); on glucose and lipid levels in diabetic and healthy volunteers: U. Smith, G. Holms, *Diabetes* (Shannon, Ire.) 45, 1 (1982); on renal function in diabetic rats: B. C. Chin *et al.*, *Biomed. Res.* 5, 273 (1982). As source of fiber in patients with non-insulin dependent diabetes: M. E. McIvor *et al.*, *Am. J. Clin. Nutr.* 40, 1 (1985). Toxicology studies: S. L. Graham *et al.*, *Toxicol. Met. Toxicol.* 19, 287 (1981). Comprehensive monograph: F. Smith, R. Montgomery, *The Chemistry of Plant and Mucilages* (Reinhold, New York, 1959) 627 pp. Goldstein *et al.* in *Industrial Gums*, R. L. Whistler (Academic Press, New York, 2nd ed., 1973) p 303.

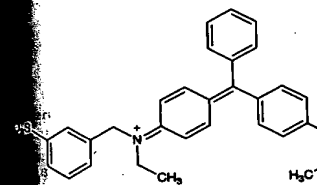
Free flowing powder. Completely sol in cold water; practically insol in oils, greases, hydrocarbons, esters. Water solns are tasteless, odorless, and of a pale, translucent gray color, and neutral. Stable to heat. Has five to eight times the thickening power of water solns may be converted to a gel by small amounts of borax. Aq dispersions are neutral. Cf. "A Comprehensive Study of Commercially Available Guar Gums" by Schlakman, A. J. Bartilucci, *Drug Standards* 25, 1 (1957). LD₅₀ in male, female rats (g/kg): 7.35, 6.9 (Graham).

USE: In paper sizing; as a protective colloid; as thickening and film forming agent for cheese, salad dressings, ice cream, soups; as a binding and disintegrating agent in tablet formulations; in pharmaceutical jelly formulations in suspensions, emulsions, lotions, creams, toothpastes; in the mining industry as a flocculant, as a filtering aid, water treatment as a coagulant aid.

THERAP CAT: Adjunct to diet, insulin or oral hypoglycemics in control of diabetes.

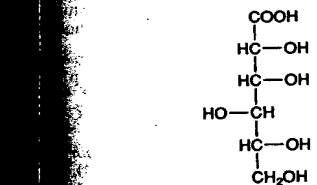
4603. Guinea Green B. *N-Ethyl-N-[4-[(4-ethoxyphenyl)methyl]amino]phenyl]phenylmethylethylamine*.

N-Ethyl-N-[4-[(4-ethoxyphenyl)methyl]amino]phenyl]phenylmethylethylamine; sodium salt; C.I. Acid Green 3; C.I. Acid Green 1; C.I. 42085. $C_{27}H_{35}N_3Na$; mol wt 433.33, H 5.11%, N 4.06%, Na 90.83%. Prepn: Jones *et al.*, *J. Assoc. Chem. Technol.* 1, 97, 30 (1964); W. H. Hanse, *Chem. Ber.* 4, 389 (1966). See also: *Chem. Ber.* 1971 p 4385.



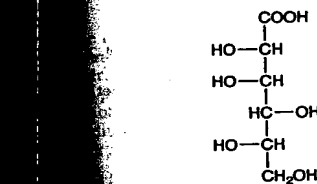
Dark green powder, or a bright green soln in water to a green soln which becomes colorless on addition of HCl and blackish-green on addition of NaOH; decolorizes the soln. Sparingly sol in concd H_2SO_4 to a yellow soln; with water, turns first yellowish-brown, then colorless. Limited use as a dye for silk and cotton. Delisted by FDA in 1971 and cosmetics.

4604. D-Gulonic Acid. $C_6H_{12}O_7$; mol wt 176.16. C 36.74%, H 6.17%, O 57.09%. Prepn: Oxidation of sodium glucuronate with chromic acid; Fischer, *Piloty, Ber.* 4, 389 (1966). See also: *Chem. Ber.* 1971 p 4385.



mp 6° (10 min) → -38.6° (10 min). The lactone spontaneously. Prepared from aq soln by alc. salt, $Ca(C_6H_{11}O_7)_2$. $[\alpha]_D^{25}$ -1.7° (c = 10) (aq soln by alc).

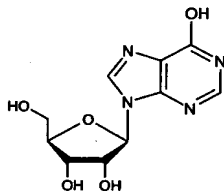
4605. D-Gulonic Acid. Xylose derivative. Mol wt 196.16. C 36.74%, H 6.17%, O 57.09%. Prepn: Fischer, Stahel, *Ber.* 24, 529 (1957). See also: *Chem. Ber.* 1971 p 4385.



mp 6° (10 min) → -38.6° (10 min). The lactone spontaneously. Prepared from aq soln by alc. salt, $[\alpha]_D^{25} +12.7^\circ$ (c = 9). **4606. D-Gulonic Acid.** $C_6H_{12}O_7$; mol wt 176.16. C 36.74%, H 6.17%, O 57.09%. Prepn: Fischer, Stahel, *Ber.* 24, 529 (1957). See also: *Chem. Ber.* 1971 p 4385.

1986) pp 408-418. Reviews: F. H. de Jong, *Oxford Rev. Reprod. Biol.* 9, 1-53 (1987); N. Ling et al., *Vitam. Horm.* (New York) 44, 1-46 (1988).

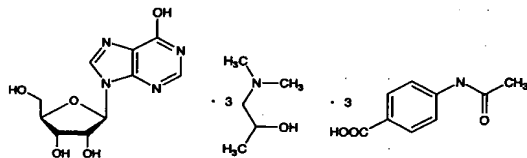
5005. Inosine. Hypoxanthine riboside; 9- β -D-ribofuranosylhypoxanthine; hypoxanthosine; Inosie; Oxiamine; Ribonosine; Trophicardyl. $C_{10}H_{12}N_4O_5$; mol wt 268.23. C 44.78%, H 4.51%, N 20.89%, O 29.82%. In meat and meat extracts, in sugar beets. Prep'd from adenosine by incubation with purified adenosine deaminase from intestine: Kalckar, *J. Biol. Chem.* 167, 445 (1947); also by the action of sodium nitrite and acetic acid on adenosine: Levene, Jacobs, *Ber.* 43, 3161 (1910); by the use of barium nitrite and H_2SO_4 : Reiff et al., U.S. pat. 3,049,536 (1962 to Zellstoff-Fabrik Waldhof). Fermentation method: Motozaki et al., U.S. pat. 3,111,459 (1964 to Ajinomoto). Structure: Brederick, *Ber.* 66, 198 (1933); *Z. Physiol. Chem.* 223, 61 (1934); Gulland, Holiday, *J. Chem. Soc.* 1936, 765.



Dihydrate, long rectangular plates from water, mp 90°. Anhydrous needles from 80% alc, dec 218° (rapid heating). $[\alpha]_D^{20} -49.2^\circ$ (c = 0.9 in H_2O). $[\alpha]_D^{20} -73^\circ$ (0.5 g + 2 ml N NaOH + 3 ml H_2O). 100 ml of the sat'd water soln at 20° contain 1.6 g inosine. Absorption spectrum: Kalckar, *loc. cit.* uv max (pH 6.0): 248.5 nm (ϵ 12200). Boiling with 0.1N H_2SO_4 yields hypoxanthin and D-ribose.

THERAP CAT: Activates cellular functions.

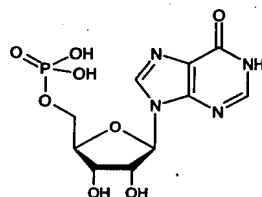
5006. Inosine Pranobex. *Inosine monof[4-(acetylaminobenzoate)] (salt) comp'd with 1-(dimethylamino)-2-propanol* (1:3); inosine:dimethylaminoisopropanol acetamidobenzoate (1:3); inosiplex; methisoprinol; NP-113; NPT-10381; Aviral; Delimmun; Immunoviral; Isoprinosin; Isoprinosina; Isoprinosine; Isoviril; Modimmunol; Pranosina; Viruxan. $C_{27}H_{36}N_{10}O_{17}$; mol wt 1115.25. C 56.00%, H 7.05%, N 12.56%, O 24.39%. Immunostimulant complex formed from the *p*-acetamidobenzoate salt of dimethylaminoisopropanol and inosine in a 3:1 molar ratio. Prep'n: P. Gordon, *Ger. pat.* 1,965,431; *idem*, U.S. pat. 3,646,007 (1971, 1972 both to Newport Pharm.). Antiviral activity: E. R. Brown, P. Gordon, *Can. J. Microbiol.* 18, 1463 (1972); R. L. Muldoon et al., *Antimicrob. Ag. Chemother.* 2, 224 (1972). Stimulatory effect on T-cell function: L. Binderup, *Int. J. Immunopharmacol.* 7, 93 (1985). Pharmacology and therapeutic potential: D. M. Campoli-Richards et al., *Drugs* 32, 383 (1986). Clinical immunopharmacology: A. J. Glasky, J. F. Gordon, *Cancer Detect. Prev. Suppl.* 1, 597 (1987). Clinical trial in subacute sclerosing panencephalitis (SSPE): C. E. Jones et al., *Lancet* 1, 1034 (1982); G. Gascon et al., *Brain Devel.* 15, 346 (1993). Clinical trial in pre-AIDS patients: C. Pedersen et al., *N. Engl. J. Med.* 322, 1757 (1990). Review of efficacy in HIV infection: C. De Simone et al., *Int. J. Immunopharmacol.* 13, Suppl. 1, 19-27 (1991).



Neutral water-soluble solid. LD₅₀ in mice and rats (mg/kg): >4000 orally and i.p. (Gordon). THERAP CAT: Immunomodulator; antiviral.

5007. Inosinic Acid. 5'-Inosinic acid; 5-inosinic acid; muscle inosinic acid; t-inosinic acid; hypoxanthine riboside-

5-phosphoric acid; IMP. $C_{10}H_{13}N_4O_8P$; mol wt 348.23. C 34.49%, H 3.76%, N 16.09%, O 36.76%, P 8.90%. Prep'd from meat extract: Levene, Bass, *Nucleic Acids* (New York) 1931) p 229; from dried sardines: Yoshida, Kageyama, Japan, pat. 732(56) (to Ajinomoto), C.A. 51, 3870b (1931). Structure: Levene, Bass, *op. cit.*, pp 187-192; Brederick, *Ber.* 66, 198 (1933); Levene, Tipson, *J. Biol. Chem.* 111, 111 (1935). Also prep'd from muscle by enzymatic deamination of muscle adenylic acid: Ostern, *Biochem. Z.* 254, 111 (1932); by hydrolysis of inosine triphosphate: Klein, *Biochem. J.* 36, 729 (1942). Studies on the enzymatic thesis: Greenberg, *J. Biol. Chem.* 190, 611 (1951); Kono, *ibid.* 217, 875 (1955). Microbial fermentation method using mutant strains of *Micrococcus glutamicus*: Kono et al., U.S. pat. 3,232,844 (1966 to Kyowa).



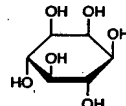
Syrup, solidifies to a glass when dried over P_2O_5 . Agreeable sour taste. $pK_1 = 2.4$; $pK_2 = 6.4$. Absorption spectrum: Kalckar, *J. Biol. Chem.* 167, 445 (1947). Sol in water, in formic acid; very sparingly sol in alcohol, ether. On boiling with acid hydrolyzes to 1 mol H_2O mol hypoxanthine, 1 mol D-ribose.

Disodium salt dihydrate, $C_{10}H_{11}N_4Na_2O_8P \cdot 2H_2O$, sol in alcohol, ether, acetone; soly in water at 20° about 1 g/100 ml. Kawasaki, *New Food Ind. (Tokyo)* 3, no. 1 (1961).

Barium salt, $C_{10}H_{11}BaN_4O_8P$. Hemipentadecahydrate, lustrous leaflets. Becomes anhydrous at 100° in *vacuo*. mp -18.5° (0.3 g of anhydrous Ba salt in 10 ml of 2.5% HCl).

USE: Its salts as flavor intensifiers, like sodium glutamate or other salts: Toi et al., U.S. pat. 3,109,741 (1964 to Ajinomoto).

5008. Inositol. *myo-Inositol*; *meso-inositol*; *D-inositol*; hexahydroxycyclohexane; cyclohexanehexol; cyclohexanehexol; inositol; mesoinositol; phaseomannite; damianite; bios I; rat antipetacted eye factor; mouse antipetacted eye factor. $C_6H_{12}O_6$; mol wt 180.16. C 40.00%, H 6.67%, O 53.28%. Widely distributed in plants and animals. Growth factor for animals and microorganisms. Isolin heart muscle: Scherer, *Ann.* 73, 322 (1850); from Woolley, *J. Biol. Chem.* 139, 29 (1941). Synthesis: land, Wishart, *Ber.* 47, 2082 (1914); Anderson, Wall, *Am. Chem. Soc.* 70, 2931 (1948). Obtained commercially from corn steep liquor, since inositol is present as acid in corn: Bartow, Walker, *Ind. Eng. Chem.* 30, 1938 (1938); U.S. pat. 2,112,553 (1938); Hoglan, Bartow, *J. Chem. Soc.* 62, 2397 (1940); Elkin, Meadows, U.S. 2,414,365 (1947); Brit. pat. 601,273 (1948 to Corn Refining). Nine possible stereoisomers: Seven are optically inactive or *meso*. Two optically active forms, the *D* and *L* forms, and several *cis,trans*-isomers occur naturally. The natural form is *cis-1,2,3,5-trans-4,6-cyclohexanehexol*, which is described here. Reviews: R. Beckmann, *Chem. Ber.* 86, 100 (1953); several authors in *Vitamins*, vol. 2, W. H. Sebrell, Jr., R. S. Harris, Eds., Academic Press, New York, (1954) pp 321-386; *ibid.* vol. 3, ed., (1971) pp 340-415.



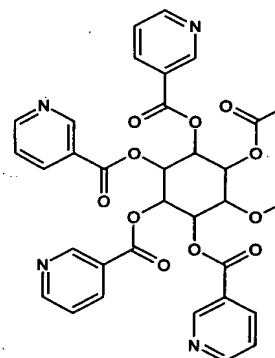
Anhydr, non-hygroscopic crystals from water, acid above 80°. Sweet taste. d 1.752. mp 225-227°.

inactive. Soly in water at 25°: 28 g/100 ml soln. Slightly sol in alcohol and other common organic solvents to litmus.

Hydrate, efflorescent crystals from water, mp 218°. Becomes anhydrous at 100° in *vacuo*. Phosphate, $C_6H_{13}O_8P$. Prep'n: *Helv. Chim. Acta* 12, 1165 (1929); *Chem. Prepn.* 2, 65 (1952). Crystals, mp 195-197°. Titrates as a dibasic acid (1 g dissolves in 3 ml H_2O). Phosphate, ether. Remarkably resistant to strong alkali. May be hydrolyzed for 14 hrs.

THERAP CAT: Vitamin B complex; lip

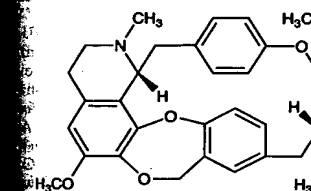
5009. Inositol Niacinate. *myo-Inositol*; hexanicotinoyl inositol; hexanicotinoyl-4,6-cyclohexane; inositol hexanicotinoyl; Dilcit; Dilxpal; Hexanid; Hexanicit; Hexopal; Lin. $C_{26}H_{38}N_2O_{11}$; mol wt 810.73. C 56.00%, H 4.51%, N 3.89%, O 23.68%. Prep'n: Badgett, *J. Soc.* 69, 2907 (1947).



Crystals, mp 254.3-254.9°. Practically insoluble.

THERAP CAT: Vasodilator (peripheral)

5010. Insularine. $C_{38}H_{40}N_2O_6$; mol wt 604.66. C 65.00%, H 6.50%, N 4.51%, O 15.46%. *Insularis* Makino and C. *oculocaryophyllaceae*. Isolin: Kondo, Yano, *J. Biol. Chem.* 185, 815 (1927); Kondo, Tomita, *At. Soc. Jpn.* 1957). Structure: Tomita, Kikuchi, *J. Soc. Jpn.* 1957).



Amorphous powder, $[\alpha]_D^{20} +28^\circ$, mp 125-126°. Absorption spectrum: Ochai, *J. Soc. Jpn.* 1929).

5011. Insulin. Polypeptide hormone secreted by beta cells that regulates carbohydrate metabolism by proteolysis from the inactive dimer composed of two polypeptide chains. Regulates carbohydrate metabolism and influences protein synthesis. Protein for which the chemical structure is determined. Also the first protein produced by recombinant